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chemical bond is a lasting attraction between atoms or ions that enables the formation of molecules and crystals the bond may result from the electrostatic force between oppositely charged ions as in ionic bonds or through the sharing of electrons as in covalent bonds the strength of chemical bonds varies considerably there are strong bonds or primary

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stepwise as in example 3 17 it is important to refrain from rounding any intermediate calculation results which can lead to rounding errors in the final result in example 3 17 the molar amount of nacl computed in the first step 1 325 mol would be properly rounded to 1 32 mol if it were to be reported however although the

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